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means that more than 300 000 l of oxygen gas are needed to produce 1 ton of sulfuric acid these numbers may give you some appreciation for the magnitude of the engineering and plumbing problems faced in industrial chemistry

web b gases that are not well mixed

c a mixture of gases that behaves as if it were a single gas

2 based on temperature the atmosphere is divided into group of answer choices

a two broad regions

b four regions ranging from the troposphere to the thermosphere

c nitrogen oxygen argon

web transcribed image text

es hot exhaust gases of an internal combustion engine are to be used to produce saturated water vapor at 2 mpa pressure the exhaust gases enter the heat exchanger at 400 c at a rate of 32 kg min while water enters at 15 c the heat exchanger is not well insulated and it is estimated that 10 percent of heat given up by

web mar 23 2021 for a given amount of a gas the volume temperature and pressure under any one set of conditions are related to the volume the temperature and the pressure under any other set of conditions by the equation

$$p_1 v_1^{\gamma} t_1^{\gamma} = p_2 v_2^{\gamma} t_2^{\gamma}$$

web chemthink behavior of gases

name per 1 what does the speed of atoms depend on describe the relationship between speed and each of these factors

i 3 what is the equation for pressure and what are the units for pressure

4 when atoms are moving in a closed container what is creating the force on the sides of the container

5 **web** figure 9 27 a two gases h₂ and o₂ are initially separated

b when the stopcock is opened they mix together the lighter gas h₂ passes through the opening faster than o₂ so just after the stopcock is opened more h₂ molecules move to the o₂ side than o₂ molecules move to the h₂ side

c after a short time both the slower moving o₂

web 1 why can't all of the gas particles in a container have very high or very low speeds because as particles collide they both increase their kinetic energy because as particles collide a gain

web gas is the state of matter in which the particles are far apart fast moving and not organised in any particular way gases are substances that exist in the gaseous state which is one of the three fundamental states of matter gases are highly compressible and feature very large intermolecular distances

web answers because an ideal gas is theoretical and hence not real because all ideal gases are invisible because an ideal gas is composed of molecules that cannot be seen with the naked eye

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